

Synthesis and antimicrobial activity of new adamantane derivatives II

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Abstract

A series of new derivatives of adamantane was synthesised. The new compound 4-(adamant-1-ylethylenoxycarbonyl)phthalanhydride obtained from 1-adamantaneethanol and trimellitic anhydride chloride, as well as 4-(adamant-1-ylmethylenoxycarbonyl)phthalanhydride, appeared very useful for preparation of a number of N-substituted phthalimides. Antimicrobial activity of newly obtained derivatives such as, for example, 4-(adamant-1-ylethylenoxycarbonyl)-N-(L-phenylalanyl)phthalimide or 4-(adamant-1-ylmethylenoxycarbonyl)-N-(L-leucyl)-phthalimide was tested against *Staphylococcus aureus*, *Bacillus* sp., *Micrococcus flavus* and *Enterococcus faecium*. The minimal inhibitory concentrations for these compounds against *Bacillus cereus* were 15 and 8 µg/ml, respectively. © 2000 Elsevier Science S.A. All rights reserved.

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1. Introduction

Many adamantane derivatives have interesting biological properties. The most known of clinical use is the antiviral drug 1-aminoadamantane (amantidine) [1]. Amantidine and many similar compounds with three-dimensional 'box-like' adamantane structure are employed in the treatment of certain neurological disorders [2]. The combination of such aliphatic core with phthalimide ring gives biologically active compounds. For instance, N-adamantylphthalimide induces tumour necrosis factor (TNF- α) in human leukaemia HL-60 cells [3], while 1-adamantylmaleinimide and its derivatives, show anticancer activity in mice and inhibition of herpes simplex virus in vitro [4,5]. Moreover, these compounds were tested against some micro-organisms such as *Staphylococcus aureus*, *Escherichia coli*, *Pseudomonas aeruginosa* and others, however their minimal inhibitory concentration (MIC) were rather high (> 100 µg/ml). Adamantaneketoxime ethers and their derivatives were tested as antibacterial and antifungus agents but mostly were found inactive (MIC > 200 µg/ml) [6].

In our previous publication we have found that a number of 4-(adamant-1-yl-methylenoxycarbonyl)-N-substituted-phthalimides showed a very strong antibacterial activity, especially when as N-substituent, alanine, phenylalanine and some ω -aminoacids were used [7]. Minimal inhibitory concentrations for these compounds against *S. aureus* strain and *Micrococcus flavus* were comparable with that of clinically used antibiotics (1–0.02 µg/ml).

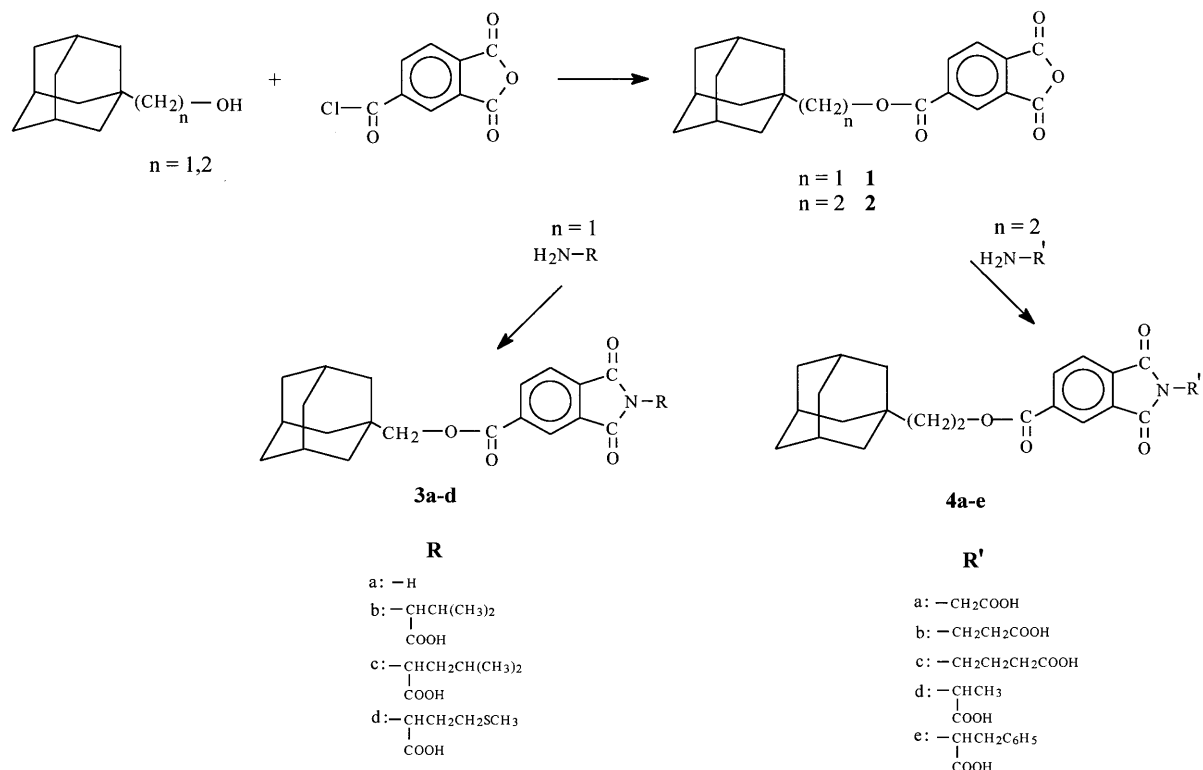
In this paper we would like to develop this line of our early studies and present two series of new phthalimide derivatives of adamantane and their antibacterial properties. The general synthetic pathway is given in Scheme 1.

2. Experimental

2.1. General methods

Melting points were taken in open capillary tubes on a Gallenkamp 5 melting point apparatus and were uncorrected. The structures of products were confirmed by elemental analysis, FTIR and ¹H NMR spectroscopy. The NMR spectra were measured on a

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Scheme 1.

Varian Gemini 200 MHz spectrometer in CDCl_3 solutions. Column flash chromatography and TLC were performed on silica gel 60 (Merck) using chloroform/methanol/acetic acid (20:1:trace) mixture as eluent. FTIR spectra were recorded on a Perkin–Elmer 2000 apparatus using the KBr pellet method. Analyses indicated by symbols were within $\pm 0.4\%$ of theoretical values. Adamantane and phthalic acid derivatives were purchased from Aldrich. Preliminary testing of the antimicrobial activity of newly synthesised compounds was performed by disc diffusion method using Mueller–Hinton agar medium under standard conditions as described by NCCLS [8]. Sterile filter paper discs were soaked in test compounds solutions prepared in EtOH–DMSO mixture (1:1). The results were read following 18 h incubation at 37°C (for *M. flavus* at 30°C). Compounds showing distinct antimicrobial activity in the above test were next examined for MIC in liquid Mueller–Hinton medium according to the appropriate NCCLS protocol, using original stock solutions [9]. *S. aureus* NCTC 4163 and *Enterococcus faecium* ATCC 6057 were purchased from the National Institute of Hygiene, Warsaw; the other microorganisms used were from the collection of the Department of Pharmaceutical Microbiology, Medical University of Warsaw, Poland.

2.2. 4-(Adamant-1-ylmethoxy)phthalic anhydride $\text{C}_{20}\text{H}_{20}\text{O}_5$ (**1**)

The synthesis of **1** was performed from 1-adamantanemethanol and trimellitic anhydride chloride according to the method described previously [10,11]. Crude product was crystallised from benzene. Yield 42%; m.p. 158°C; FTIR (cm^{-1}): $\text{C}=\text{O}_{\text{ester}}$ 1731, $\text{C}=\text{O}_{\text{anh}}$ 1785 and 1857.

2.3. 4-(Adamant-1-ylethoxy)phthalic anhydride $\text{C}_{21}\text{H}_{22}\text{O}_5$ (**2**)

The synthesis of **2** was performed from 1-adamantanemethanol and trimellitic anhydride chloride according to method described previously [10,11]. Crude product was crystallised from a toluene–hexane mixture. Yield 49%; m.p. 133–135°C; FTIR (cm^{-1}): $\text{C}=\text{O}_{\text{ester}}$ 1726, $\text{C}=\text{O}_{\text{anh}}$ 1785 and 1855.

2.4. 4-(Adamant-1-ylmethoxy)phthalimide $\text{C}_{20}\text{H}_{21}\text{NO}_4$ (**3a**)

A total of 1.36 g (5 mmol) of **1** and 0.15 g (2.5 mmol) of urea were melted and the reaction was run over 10 min at 250°C. Crude product was crystallised from a

benzene–methanol mixture. Yield 78%; m.p. 214°C. $R_f = 0.21$; FTIR (cm^{-1}): N–H 3275, C=O_{imide} 1779 and 1753, C=O_{ester} 1726; $^1\text{H NMR } \delta$ (ppm): 1.63–2.03 (m, H_{adamantane}), 3.99 (s, CH₂O), 7.94–8.50 (m, H_{arom}), 7.98 (s, N–H).

2.5. 4-(Adamant-1-ylmethylenoxycarbonyl)-*N*-substituted-phthalimides (**3b–d**)

A total of 1.36 g (5 mmol) of **1** and the proper amino acid were refluxed in dry DMF for 3 h. The mixtures were poured into diluted HCl (1%). Crude products were filtered off and purified by column chromatography (silica gel) using chloroform/methanol/acetic acid (v/v, 25:1:traces) as eluent.

2.6. 4-(Adamant-1-ylmethylenoxycarbonyl)-*N*-(*L*-valinyl)phthalimide C₂₅H₂₉NO₆ (**3b**)

M.p. 140–142°C, 1.33 g, 65%; $R_f = 0.33$; FTIR (cm^{-1}): C=O_{imide} 1780 and 1725; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 0.91–1.18 (d, CH₃), 1.63–2.02 (m, H_{adamantane}), 3.98 (s, CH₂O), 4.65 (d, N–CH), 7.93–8.48 (m, H_{arom}).

2.7. 4-(Adamant-1-ylmethylenoxycarbonyl)-*N*-(*L*-leucyl)phthalimide C₂₆H₃₁NO₆ (**3c**)

M.p. 99°C, 0.99 g, 57%; $R_f = 0.35$; FTIR (cm^{-1}): C=O_{imide} 1779 and 1724; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 0.87–0.94 (m, CH_{leucine}), 1.62–2.02 (m, H_{adamantane}), 2.80 (t, CH₂C=O), 3.98 (s, CH₂O), 4.92 (m, N–CH), 7.90–8.45 (m, H_{arom}).

2.8. 4-(Adamant-1-ylmethylenoxycarbonyl)-*N*-(*L*-methionyl)phthalimide C₂₅H₂₉NO₆S (**3d**)

M.p. 56–58°C, 0.90 g, 56%; $R_f = 0.38$; FTIR (cm^{-1}): C=O_{imide} 1780 and 1724; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.63–1.73 (m, H_{adamantane}), 2.04 (s, CH₃S), 2.48 (m, –CH₂–), 3.97 (s, CH₂CO), 5.03 (m, N–CH); 7.89–8.45 (m, H_{arom}).

2.9. 4-(Adamant-1-ylethylenoxycarbonyl)-*N*-substituted phthalimides (**4a–e**)

A total of 1.36 g (5 mmol) of **2** and the proper amino acid were refluxed in dry DMF for 3 h. The mixtures were poured into diluted HCl (1%). Crude products **4a–c** were filtered off and crystallised from ethanol; crude products **4d** and **4e** were purified by column chromatography (silica gel) using chloroform/methanol/acetic acid (v/v, 25:1:traces) as eluent.

2.10. 4-(Adamant-1-ylethylenoxycarbonyl)-*N*-glycylphthalimide C₂₃H₂₅NO₆ (**4a**)

M.p. 103°C, 1.23 g, 72%; $R_f = 0.37$; FTIR (cm^{-1}): C=O_{imide} 1778 and 1720; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.56–2.02 (m, H_{adamantane}), 4.40 (t, CH₂O), 4.51 (s, N–CH₂), 7.94–8.51 (m, H_{arom}).

2.11. 4-(Adamant-1-ylmethylenoxycarbonyl)-*N*-(2-carboxydimethylene)phthalimide C₂₄H₂₇NO₆ (**4b**)

M.p. 181°C, 1.44 g, 76%; $R_f = 0.37$; FTIR (cm^{-1}): C=O_{imide} 1778 and 1720; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.56–2.02 (m, H_{adamantane}), 2.82 (t, CH₂COO), 4.03 (t, CH₂O), 4.44 (t, N–CH₂), 7.90–8.48 (m, H_{arom}).

2.12. 4-(Adamant-1-ylethylenoxycarbonyl)-*N*-(3-carboxytrimethylene)phthalimide C₂₅H₂₉NO₆ (**4c**)

M.p. 163°C, 1.33 g, 75%; $R_f = 0.41$; FTIR (cm^{-1}): C=O_{imide} 1776 and 1720; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.56–2.02 (m, H_{adamantane} and CH₂), 2.43 (t, CH₂COO), 3.79 (s, CH₂O), 4.33 (t, N–CH), 7.93–8.46 (m, H_{arom}).

2.13. 4-(Adamant-1-ylethylenoxycarbonyl)-*N*-(*L*-alanyl)phthalimide C₂₄H₂₇NO₆ (**4d**)

M.p. 84°C, 1.00 g, 55%; $R_f = 0.37$; FTIR (cm^{-1}): C=O_{imide} 1779 and 1724; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.55–1.97 (m, H_{adamantane} and CH), 4.43 (t, CH₂O), 4.98 (q, N–CH), 7.87–8.44 (m, H_{arom}).

2.14. 4-(Adamant-1-ylethylenoxycarbonyl)-*N*-(*L*-phenylalanyl)phthalimide C₃₀H₃₁NO₆ (**4e**)

M.p. 72–74°C, 1.11 g, 57%; $R_f = 0.35$; FTIR (cm^{-1}): C=O_{imide} 1780 and 1727; $^1\text{H NMR (CDCl}_3)$ δ (ppm): 1.57–1.97 (m, H_{adamantane}), 3.61 (m, CH₂–phenyl), 4.41 (t, CH₂O), 5.22 (m, N–CH), 7.17 (s, H_{phenylala}), 7.37–8.35 (m, H_{arom}).

3. Results and discussion

The trimellitimidates obtained from **1** and glycine, 4-aminobutyric acid, β -alanine, L-alanine and L-phenylalanine described previously, exhibited strong antimicrobial activity [7]. These findings encouraged us to develop this series of compounds. Unsubstituted imide **3a**, as well as its derivatives synthesised from L-valine, L-leucine and L-methionine **3b–d**, respectively, were first testing by the agar disc-diffusion method using the Gram-positive strains of *S. aureus*, *M. flavus*, and certain strains of *Bacillus* (Table 1). The Gram-negative bacteria: *Bordetella bronchiseptica*, *Pseudomonas aerug-*

Table 1
Antibacterial in vitro activity expressed as diameter of growth inhibitory area

Bacteria strain	Diameter of growth inhibitory area (mm)								
	Compound ^a								
	3b ^b	3c	3d	4a	4b	4c	4d	4e	Control
<i>Staphylococcus aureus</i> ATCC 25923	20	20	22	11		12	11	15	24 ^c
<i>Staphylococcus aureus</i> NTCC 4163	24	22	23			11	18	28	22
<i>Staphylococcus aureus</i> ATCC 6538 P	24	25	25	10		20	12	16	22
<i>Micrococcus flavus</i> NCIB 8166	21	19	21	11	13	14	13	16	24
<i>Bacillus stearothermophilus</i>	26	19	20		11	12	12	16	16
<i>Bacillus subtilis</i> ATCC 6633	27	24	20		11	12	13	15	18
<i>Bacillus cereus</i>	24	27	18		13	14		20	16

^a 800 µg per 8 mm disc.

^b Compound **3a** was completely inactive.

^c Nitrofurantoin 300 µg per 8 mm disc.

Table 2
Antibacterial in vitro activity expressed as MIC

Bacteria strain	Minimal inhibitory concentration (µg/ml)							
	Compound							
	3b	3c	3d	4a	4b	4c	4d	4e
<i>Staphylococcus aureus</i> ATCC 25923	100	20	100	>100	>100	100	100	30
<i>Staphylococcus aureus</i> NTCC 4163	100	15	25	>100	100	20	30	20
<i>Staphylococcus aureus</i> ATCC 6538 P	50	15	25	>100	100	30	50	20
<i>Micrococcus flavus</i> NCIB 8166	80	80	100	>100	>100	>100	>100	20
<i>Bacillus stearothermophilus</i>	50	25	100	>100	>100	100	75	30
<i>Bacillus subtilis</i> ATCC 6633	50	50	100	>100	>100	80	75	30
<i>Bacillus cereus</i>	50	8	80	>100	>100	80	75	15

inosa and strains belonging to the family *Enterobacteriaceae* were resistant to all testing compounds.

As one can see from Table 1, for Gram-positive bacteria, the diameters of growth inhibition area of compounds studied were in the range 11–27 mm. Noteworthy, for compounds tested previously by us, the greatest diameters were 45–47 mm (for the *Bacillus* strain).

Next, the MIC of the most active compounds was determined in liquid Mueller–Hinton medium. As can be seen from Table 2, **3c** is the most effective agent and shows a particularly strong antimicrobial activity against *Bacillus cereus* and *Staphylococcus* strains.

The trimellitimidides obtained from **2** make a new series of compounds (**4a–e**). For their synthesis we have chosen, among others, glycine, L-alanine and L-phenylalanine. These amino acids, in the former case, gave the imides with the strongest antibacterial activity. For newly obtained esterimidides **4a–e** the diameters of growth inhibition area and MICs measured are shown in Tables 1 and 2, respectively.

It should be noted that the core structures of **3** and **4** are similar. The first series is obtained from 1-adamantanemethanol while the second is synthesised from 1-adamantaneethanol. Antibacterial studies of series **4**, as well as our earlier findings, show that adamantaneethanol derivatives are less active against microorganisms than compounds **3a–d**. We believe that the reason is the elongation of the spacer between the adamantane core and phthalimide ring in these molecules (**4a–e**).

The results shown in this paper and our former findings, point to a significant influence of amino acids used in the synthesis, on trimellitimidides' biological properties. On the other hand, L- and D-enantiomers of amino acid derivatives showed almost the same antiseptic efficacy [7]. Therefore *N*-alanyl and *N*-phenylalanyl substituents in appropriate adamantane derivatives, caused greatest antibacterial activity, while, the unsubstituted imide **3a**, as well as **4a,b** were almost inactive.

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